

# School of Chemical Engineering research and innovation NESA aligned teacher resource

Conceptual links between UNSW Chemical Engineering research & the NESA syllabus	Direct links to NESA syllabus with associated UNSW Chemical Engineering outreach activities/resources					
	Syllabus	Outcomes	Content	Related UNSW Chemical Engineering outreach activity/resources		
SHINE 3.0 & Sustainable Hydrogen from Waste Biomass - Chemical reactions; decomposition reactions - Factors that affect the rate of a chemical reaction; temperature, catalysts - Organic chemistry; biofuels - Analysis of chemical	Stage 6 Chemistry (Prelim)	explores the many different types of chemical reactions, in particular the reactivity of metals, and the factors that affect the rate of chemical reactions CH11-10	Module 3 - Chemical reactions Inquiry question: What are the products of a chemical reaction?  Students:  conduct investigations to predict and identify the products of a range of reactions, for example:  synthesis  decomposition  combustion  precipitation  acid/base reactions  acid/carbonate reactions (ACSCH042, ACSCH080)	Hydrogen Car Race Challenge		
systems and the associated implications for society and the environment; sustainability considerations			Inquiry question: What affects the rate of a chemical reaction?  Students:  analyse and report on how the rate of a chemical reaction can be affected by a range of factors, including but not limited to:  temperature  surface area of reactant(s)  concentration of reactant(s)  catalysts (ACSCH042)	Yr 11 Chemistry lab experience (catalysts)		

		analyses the energy considerations in the driving force for chemical reactions CH11-11	<ul> <li>investigate the role of activation energy, collisions and molecular orientation in collision theory</li> <li>explain a change in reaction rate using collision theory (ACSCH003, ACSCH046)</li> <li>Module 4 - Drivers of Reactions</li> <li>Energy Changes in Chemical Reactions</li> <li>Inquiry question: What energy changes occur in chemical reactions?</li> <li>model and analyse the role of catalysts in reactions (ACSCH073)</li> </ul>	Yr 11 Chemistry lab experience (catalysts)
	Stage 6 Chemistry (HSC)	analyses the structure of, and predicts reactions involving, carbon compounds CH12-14	Alcohols Inquiry question: How can alcohols be produced and what are their properties?  compare and contrast fuels from organic sources to biofuels, including ethanol  Reactions of Organic Acids and Bases Inquiry question: What are the properties of organic acids and bases?  draft and construct flow charts to show reaction pathways for chemical synthesis, including those that involve more than one step	
[ViXIVA]		describes and evaluates chemical systems used to design and analyse chemical processes CH12-15	Module 8 - Applying chemical ideas  Chemical Synthesis and Design Inquiry question: What are the implications for society of chemical synthesis and design?	



			Students:  evaluate the factors that need to be considered when designing a chemical synthesis process, including but not limited to:  availability of reagents  reaction conditions (ACSCH133)  yield and purity (ACSCH134)  industrial uses (eg pharmaceutical, cosmetics, cleaning products, fuels) (ACSCH131)  environmental, social and economic issues	Watch video: UNSW SHINE 3.0 with Scientia Professor Rose Amal
	Stage 5 Science (new syllabus)	evaluates current and alternative energy use based on ethical and sustainability considerations SC5-EGY-01	Energy  Sources of Energy Students:  - Identify different types of energy sources - Describe how electrical energy can be produced from different types of sources - Evaluate the advantages and disadvantages of using renewable and non-renewable sources of energy to generate electricity, including efficiency, economical and technological considerations	
			Global future energy needs Students:  - Evaluate ways to optimise current energy use - Examine data to identify past trends in energy use, and predict possible future demands, at a state, national and global level - Explain reasons for the development of alternative sources of energy	
[1000]		analyses the impact of human activity on the natural world <b>SC5-ENV-01</b>	Environmental Sustainability  Sustainability Students:  - Identify the principles and goals of sustainability - Apply scientific understanding to propose valid solutions to identified problems relating to sustainability	



describes a range of reaction types SC5-RXN-01

#### Reactions

## Reactions in context Students:

 Investigate a chemical or nuclear reaction used in industry to produce an important product



#### Read article:

New 'artificial leaf' can create ammonia directly from pure sunlight

### **Related UNSW degrees:**

**Bachelor of Engineering (Honours)(Chemical)** 

**Bachelor of Engieering (Honours) (Chemical Product Engineering)** 

**Bachelor of Engineering (Honours) (Renewable Energy)** 

Bachelor of Engineering (Honours) (Geoenergy & Geostorage)

#### Related career fields:

- Pharmaceuticals
- Food and Beverage
- Chemical Processing
- Energy Generation and Storage
- Environmental Services
- Minerals

For more information on this and other activities within the School of Chemical Engineering, please follow the links:

UNSW School of Chemical Engineering - <a href="https://www.unsw.edu.au/engineering/our-schools/chemical-engineering">https://www.unsw.edu.au/engineering/our-schools/chemical-engineering</a>

Particles and Catalysis Research Group - <a href="https://www.pcrg.unsw.edu.au/">https://www.pcrg.unsw.edu.au/</a>

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